

Molten salt eutectics from atomistic alchemical simulations

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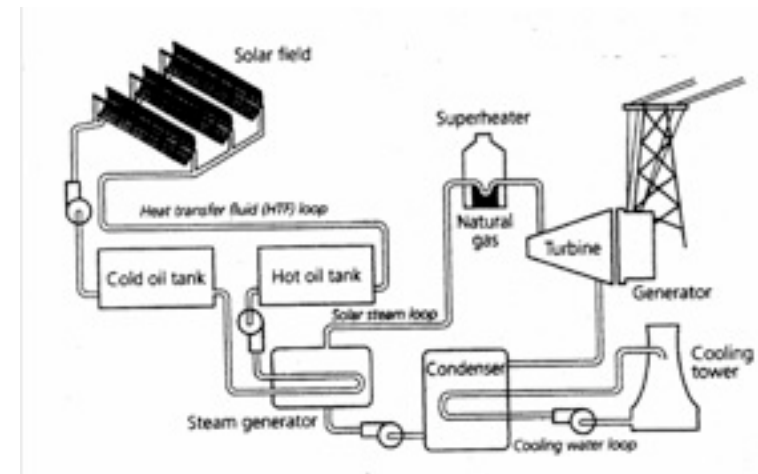
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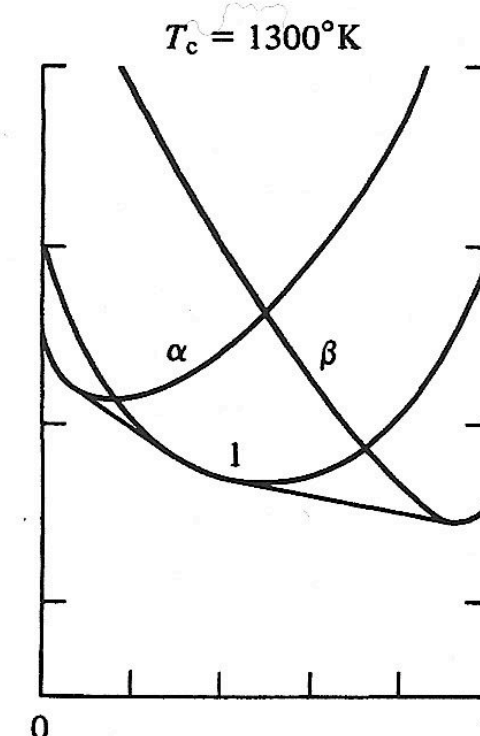
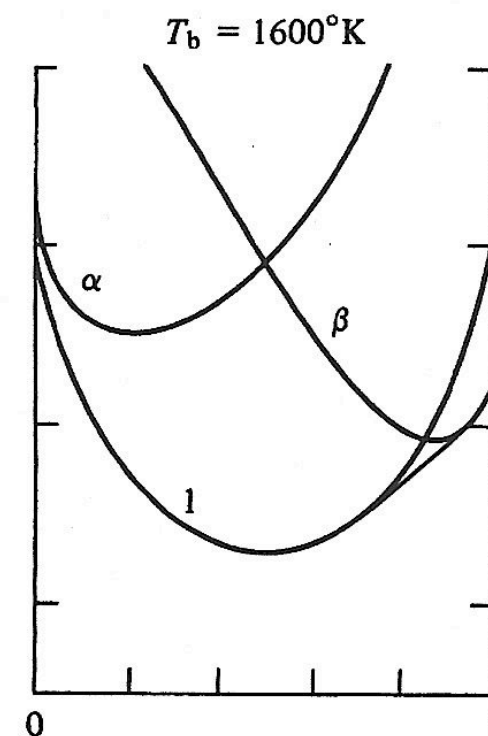
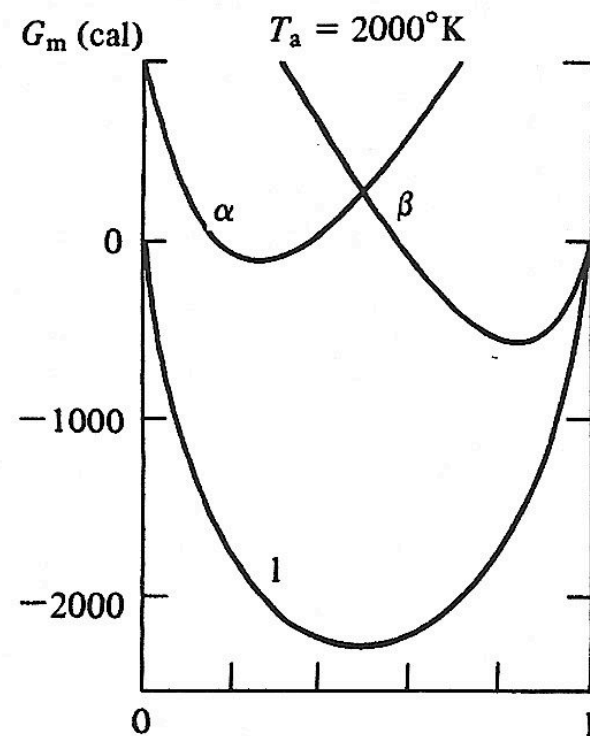
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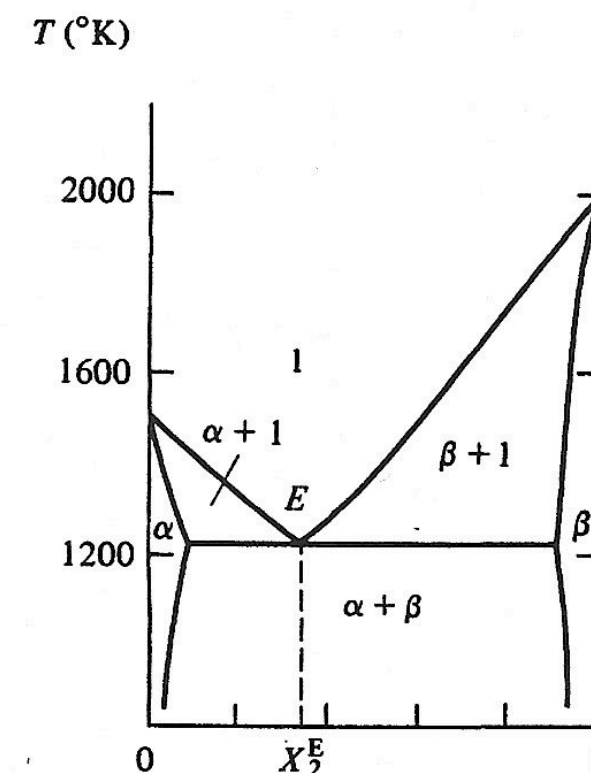
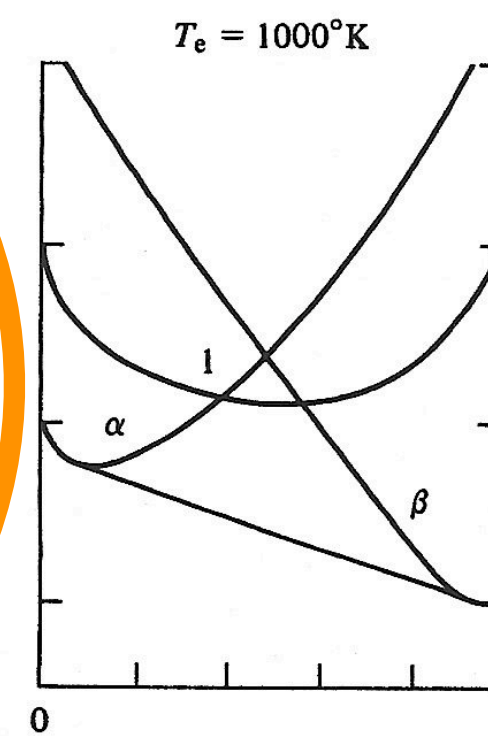
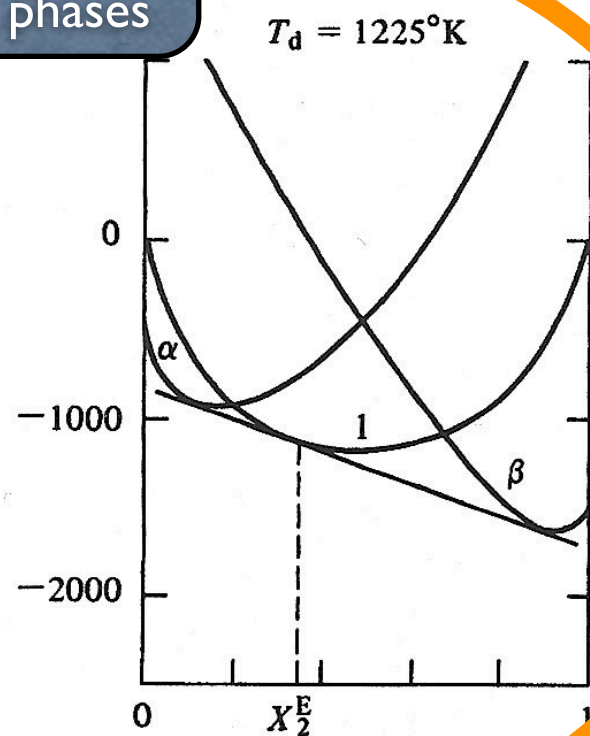


Lammps workshop
August 11, 2011





Common tangent at eutectic point between liquid and solid phases



Computing the free energy of a liquid mixture

$$G_{mix} = x_A G_A + x_B G_B + RT \sum_{i=A,B} x_i \ln x_i + G^{excess}$$

Free energy of
binary mixture at
T, P, x_A , x_B

Free energy
of Pure
species A at
T, P

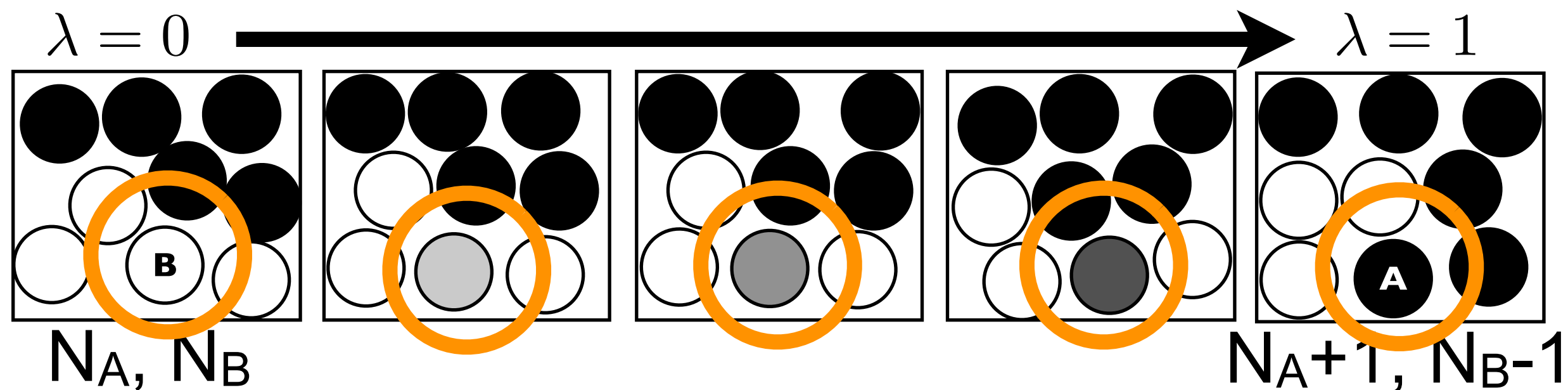
Free energy
of Pure
species B at
T, P

Free energy contribution
arising from entropy of mixing

Residual contribution
from energetic
interaction between
molecules of A and B.

Alchemical Transformation

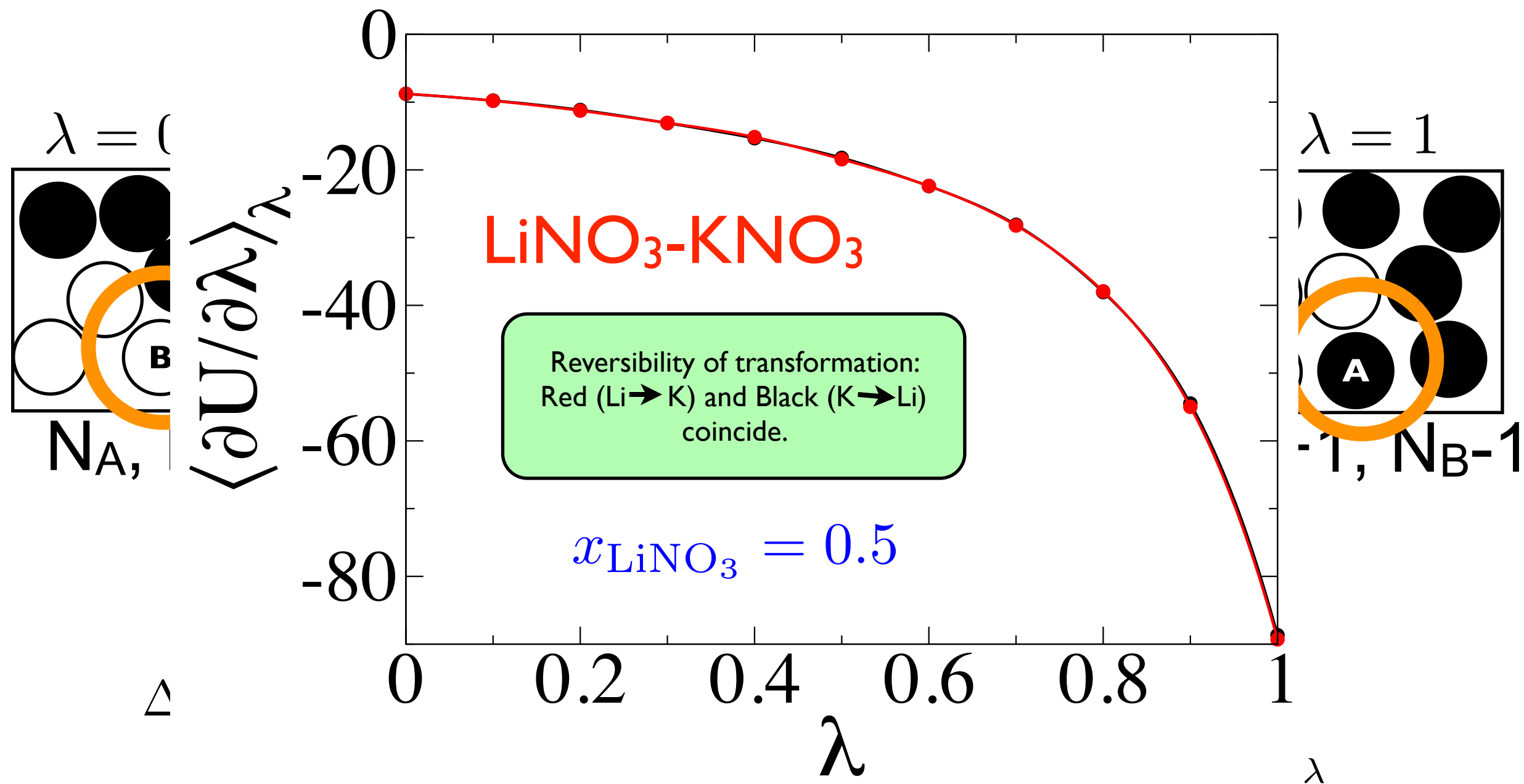
$$\frac{\partial G^{excess}}{\partial x} \approx \frac{\Delta G}{\Delta x}$$



$$u_{ij}(\lambda) = \lambda u_{iA} + (1 - \lambda) u_{iB}$$

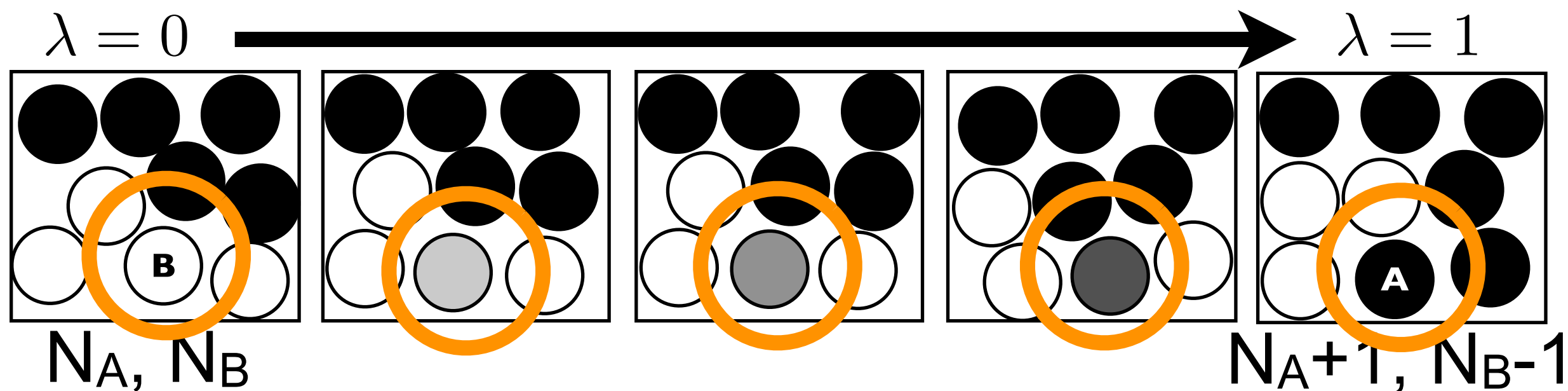
$$\Delta\mu_{AB} = \int_0^1 d\lambda \left\langle \frac{\partial U}{\partial \lambda} \right\rangle_\lambda = \int_0^1 d\lambda \left\langle \sum_i (u_{iA} - u_{iB}) \right\rangle_\lambda$$

Alchemical Transformation



Alchemical Transformation

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fix 1 all adapt 1 pair lj/cut epsilon 5 * v_lj scale yes
compute 1 all ti lj/cut 5 v_lj v_dlj tail v_lj v_dlj
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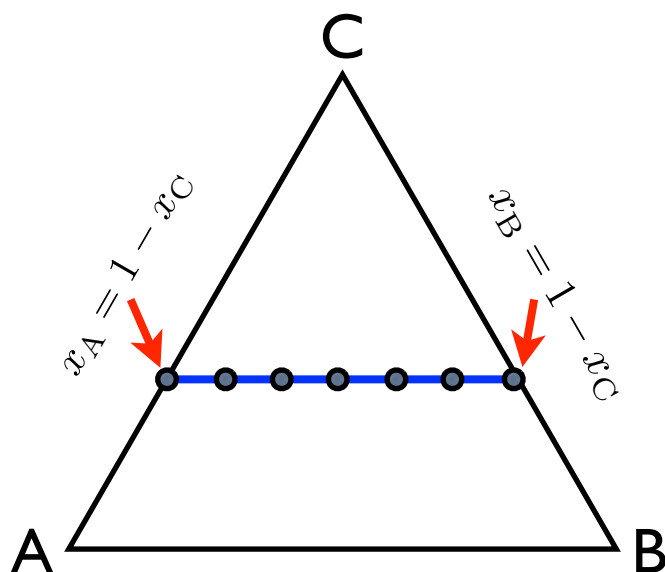

Excess free energies from $\Delta\mu_{AB}$

Binary:

$$g_{\ell}^{ex}(x_A, x_B) = \int_0^{x_A} dx'_A \Delta\mu_{AB}(x'_A) - x_A \int_0^1 dx'_A \Delta\mu_{AB}(x'_A)$$

Ternary:

$$g_{\ell}^{ex}(x_A, x_B, x_C) = \int_0^{x_A} dx'_A \Delta\mu_{AB}(x'_A, x_C) - \frac{x_A}{x_A + x_B} \int_0^{x_A + x_B} dx'_A \Delta\mu_{AB}(x'_A, x_C) + \frac{x_A}{x_A + x_B} g_{AC\ell}^{ex}(x_C) + \frac{x_B}{x_A + x_B} g_{BC\ell}^{ex}(x_C)$$

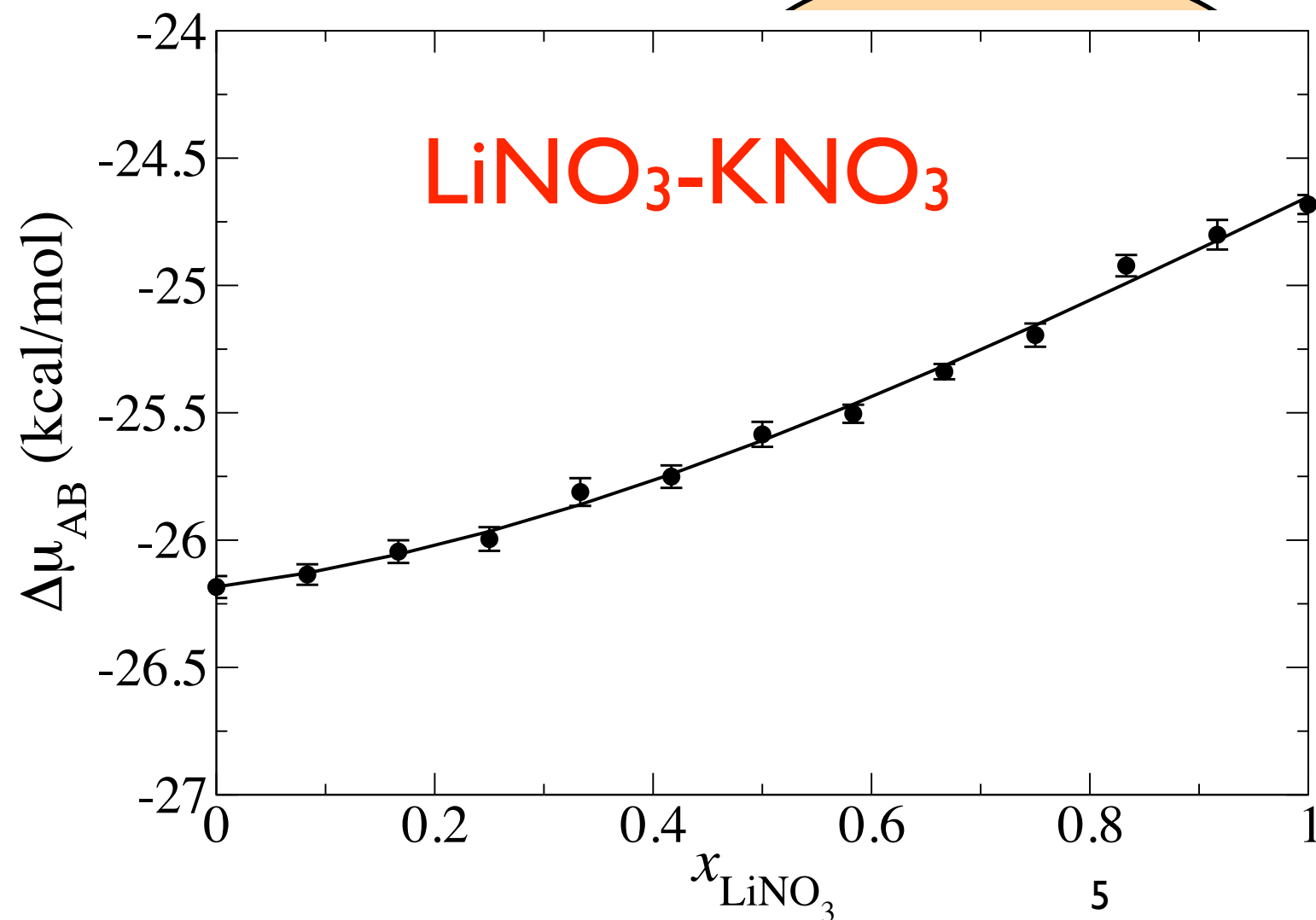


At each x_C , alchemical changes conducted at x_A mesh points

Excess free energies from $\Delta\mu_{AB}$

Binary:

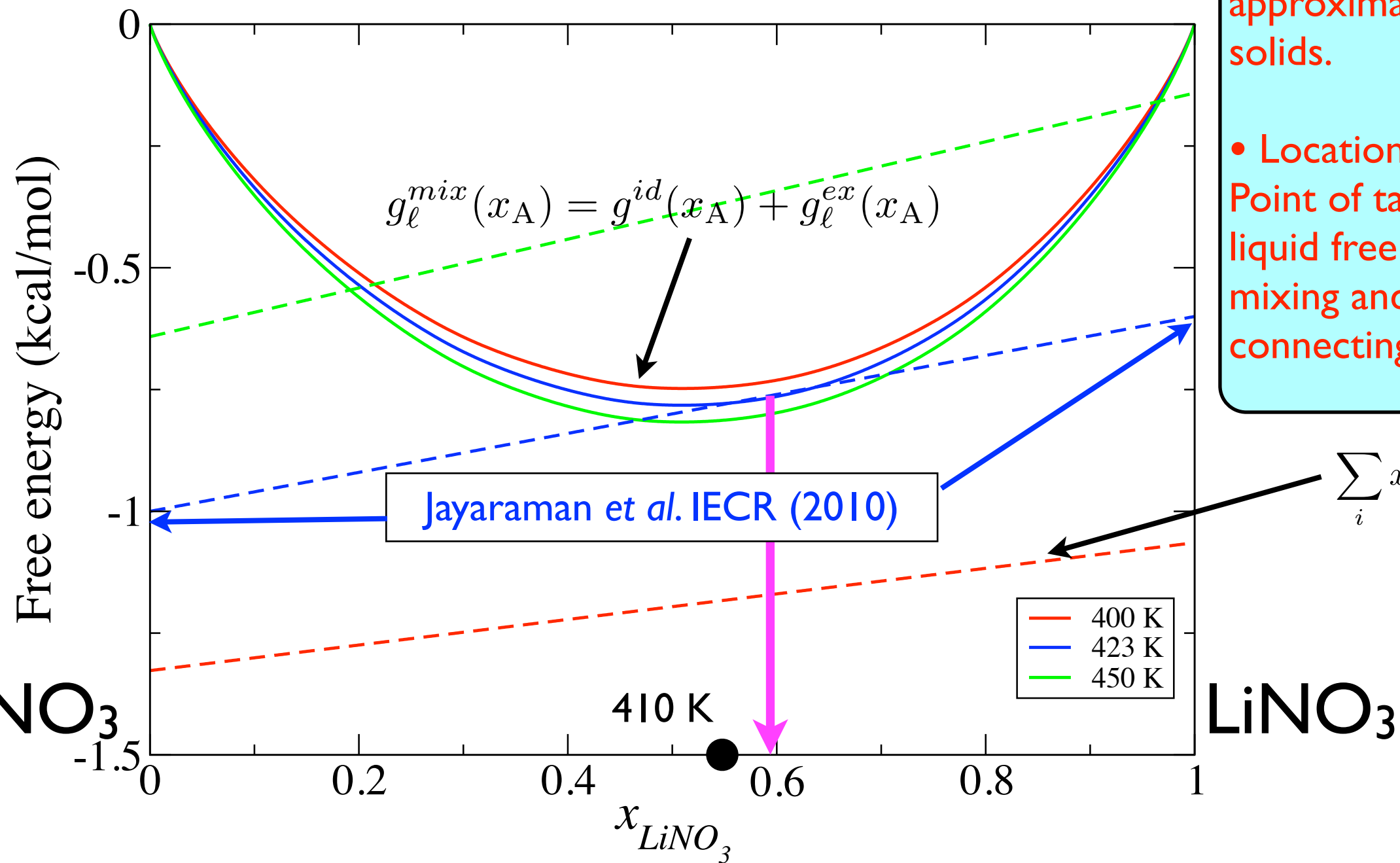
$$g_{\ell}^{ex}(x_A, x_B) = \int_0^{x_A} dx'_A \Delta\mu_{AB}(x'_A) - x_A \int_0^1 dx'_A \Delta\mu_{AB}(x'_A)$$



$$\frac{x_A}{x_A + x_B} \int_0^{x_A + x_B} dx'_A \Delta\mu_{AB}(x'_A, x_C) + \frac{x_B}{x_A + x_B} g_{BCL}^{ex}(x_C)$$

changes conducted at x_A mesh points

LiNO₃-KNO₃ mixture - Free energy vs composition



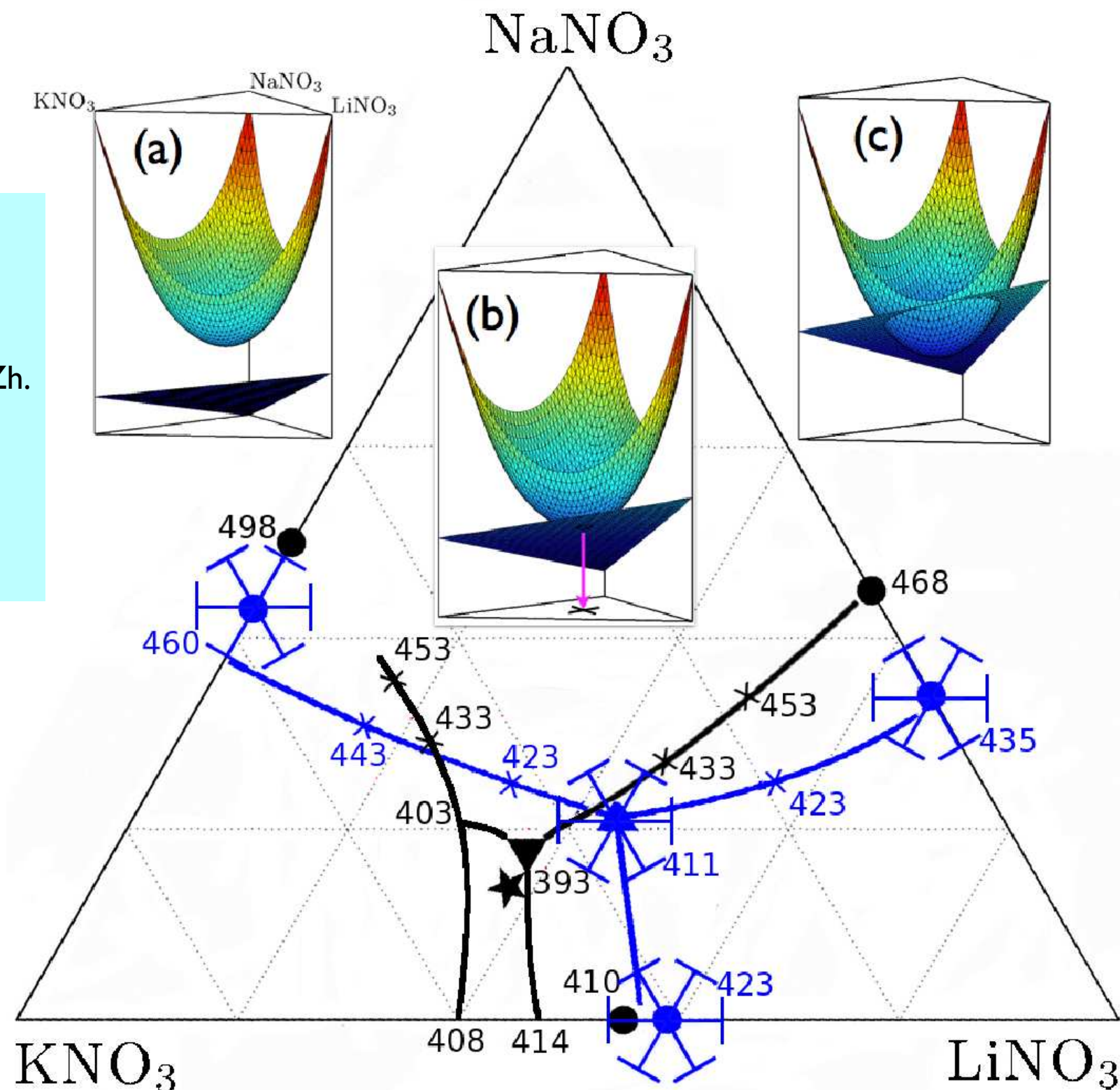
- Approximation: "Simple Eutectic" - Solid phases approximated by pure solids.
- Location of eutectic : Point of tangency between liquid free energy of mixing and solid-liquid connecting line

LiNO₃-NaNO₃-KNO₃ ternary

¹M.J. Maeso and J. Largo,
Thermochimica Acta, **223**,
145-156 (1993) - Binaries

²A.G. Bergman and K. Noguev, Zh.
Neorg. Khim., **9** 1423 (1964) -
Ternary

³H. R. Carveth, J. Phys. Chem., **2**,
209 (1898) - ★



S. Jayaraman, A. P. Thompson and O. A. von Lilienfeld, PRE rapid communication, accepted (2011)



Conclusions

- Thermodynamic integration based method developed to compute free energies of mixing from MD simulations
- Tangent method extended to compute approximations of eutectic compositions assuming a “simple eutectic”
- Need per-atom contributions from pppm!

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fix 1 all adapt 1 pair lj/cut epsilon * 5 v_lj scale yes  
compute 1 all ti lj/cut 5 v_lj v_dlj tail v_lj v_dlj
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